

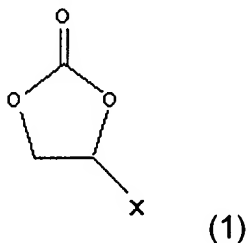
**IN THE CLAIMS:**

The text of all pending claims, (including withdrawn claims) is set forth below. Cancelled and not entered claims are indicated with claim number and status only. The claims as listed below show added text with underlining and deleted text with ~~strike through~~. The status of each claim is indicated with one of (original), (currently amended), (cancelled), (withdrawn), (new), (previously presented), or (not entered). Please AMEND claims 1, 6, 29, 37 and 38 and CANCEL claims 5 and 36, without prejudice or disclaimer of the subject matter recited therein, in accordance with the following:

1. **(CURRENTLY AMENDED)** An electrolyte of a lithium secondary battery comprising:  
 lithium salts;  
 a first organic solvent; and  
 a carbonate-based additive compound having substituents selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>),  
wherein the carbonate-based additive compound is added in an amount of 0.01 to 10 wt%.

2. **(PREVIOUSLY PRESENTED)** The electrolyte of a lithium secondary battery according to claim 1, wherein the carbonate-based additive compound is a cyclic carbonate.

3. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 1, wherein the carbonate-based additive compound is a carbonate represented by formula (1).



wherein X is selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>).

4. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 1, wherein the carbonate-based additive compound is fluoroethylene carbonate.

5. **(CANCELLED)**

6. **(CURRENTLY AMENDED)** The electrolyte of a lithium secondary battery according to ~~claim 5~~claim 1, wherein the carbonate-based additive compound is added in an amount of 0.01 to 5 wt% based on the total amount of the electrolyte.

7. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 1, wherein the lithium salts are at least one selected from the group consisting of LiPF<sub>6</sub>, LiBF<sub>4</sub>, LiSbF<sub>6</sub>, LiAsF<sub>6</sub>, LiClO<sub>4</sub>, LiCF<sub>3</sub>SO<sub>3</sub>, Li(CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>N, LiC<sub>4</sub>F<sub>9</sub>SO<sub>3</sub>, LiSbF<sub>6</sub>, LiAlO<sub>4</sub>, LiAlCl<sub>4</sub>, LiN(C<sub>x</sub>F<sub>2x+1</sub>SO<sub>2</sub>)(C<sub>y</sub>F<sub>2y+1</sub>SO<sub>2</sub>) (where x and y are natural numbers), LiCl, and LiI.

8. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 7, wherein the lithium salts are used in a concentration ranging from 0.6 to 2.0 M.

9. **(PREVIOUSLY PRESENTED)** The electrolyte of a lithium secondary battery according to claim 1, wherein the first organic solvent has a boiling point of 100°C.

10. **(PREVIOUSLY PRESENTED)** The electrolyte of a lithium secondary battery according to claim 9, wherein the first organic solvent has a boiling point of 150 °C.

11. **(PREVIOUSLY PRESENTED)** The electrolyte of a lithium secondary battery according to claim 10, wherein the first organic solvent has a boiling point of 200 °C.

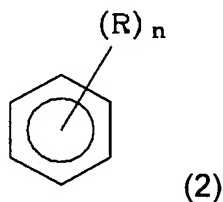
12. **(PREVIOUSLY PRESENTED)** The electrolyte of a lithium secondary battery according to claim 1, wherein the first organic solvent is selected from the group consisting of γ-butyrolactone, ethylene carbonate, dipropyl carbonate, acid anhydride, N-methyl

pyrrolidone, N-methylacetamide, N-methyl formamide, acetonitrile, dimethyl formamide, sulforane, dimethyl sulfoxide, dimethyl sulfite, and a mixture thereof.

13. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 1, wherein the electrolyte further comprises a second organic solvent.

14. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 13, wherein the second organic solvent is selected from the group consisting of dimethyl carbonate (DMC), diethyl carbonate (DEC), methylpropyl carbonate (MPC), ethylpropyl carbonate (EPC), methylethyl carbonate (MEC), propylene carbonate (PC), butylene carbonate (BC), and a mixture thereof.

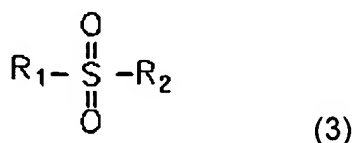
15. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 1, wherein the electrolyte further comprises an aromatic hydrocarbon organic solvent of formula (2):



wherein R is a halogen or a  $C_1$  to  $C_{10}$  alkyl, and n is an integer of 0 to 6.

16. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 15, wherein the aromatic hydrocarbon organic solvent is selected from the group consisting of benzene, fluorobenzene, toluene, fluorotoluene, trifluorotoluene, xylene, and a mixture thereof.

17. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 1, wherein the electrolyte further comprises an organic sulfone-based compound of formula (3):



wherein  $R_1$  and  $R_2$  are independently selected from the group consisting of a primary, secondary or tertiary alkyl group, an alkenyl group, an aryl group, and a cycloalkyl group, or  $R_1$  and  $R_2$  are bound together to form a cyclic ring.

18. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 17, wherein  $R_1$  and  $R_2$  are independently selected from the group consisting of a  $C_1$  to  $C_4$  alkyl, a  $C_2$  to  $C_4$  alkenyl, a  $C_6$  to  $C_{14}$  aryl, and a  $C_3$  to  $C_6$  cycloalkyl.

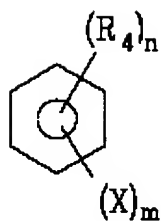
19. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 17, wherein  $R_1$  and  $R_2$  are independently a halogen-substituted alkyl group, an alkenyl group, an aryl group, or a cycloalkyl group.

20. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 17, wherein the sulfone-based compound is selected from the group consisting of vinyl sulfone, methyl sulfone, methylvinyl sulfone, ethylvinyl sulfone, phenyl sulfone, phenylvinyl sulfone, chlorophenylvinyl sulfone, fluorophenylvinyl sulfone, benzyl sulfone, tetramethylene sulfone, butadiene sulfone, and a mixture thereof.

21. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 17, wherein the sulfone-based compound is added in an amount of 0.01 to 10 wt% based on the total amount of the electrolyte.

22. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 21, wherein the sulfone-based compound is added in an amount of 0.01 to 6 wt% based on the total amount of the electrolyte.

23. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 1, wherein the electrolyte further comprises a compound of formula (4):



(4)

wherein  $R_4$  is a  $C_1$  to  $C_{10}$  alkyl, a  $C_1$  to  $C_{10}$  alkoxy, or a  $C_6$  to  $C_{10}$  aryl,  $X$  is a halogen, and  $m$  and  $n$  are integers ranging from 1 to 5, where  $m+n$  is less than or equal to 6.

24. **(WITHDRAWN)** The electrolyte of a lithium battery according to claim 23, wherein the compound of formula (4) is a compound selected from the group consisting of 3-fluoroanisole, 3-chloroanisole, 3-bromoanisole, 4-fluoroanisole, 4-chloroanisole, 4-bromoanisole, 2,4-difluoroanisole, 3,5-difluoroanisole, 3-chloro-5-fluoroanisole, and a mixture thereof.

25. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 23, wherein the compound of formula (4) is added in an amount of 0.01 to 10 wt% based on the total amount of the electrolyte.

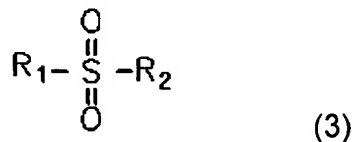
26. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 25, wherein the compound of formula (4) is added in an amount of 0.01 to 6 wt% based on the total amount of the electrolyte.

27. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 1, wherein the electrolyte further comprises a swelling-inhibiting additive.

28. **(WITHDRAWN)** The electrolyte of a lithium secondary battery according to claim 27, wherein the swelling-inhibiting additive is selected from the group consisting of propane sultone, bisphenol, dimethylfuran, 1,3-propandiol cyclic sulfate, N-acetylcaprolactam, and a mixture thereof.

29. **(CURRENTLY AMENDED)** An electrolyte of a lithium secondary battery comprising lithium salts; an organic solvent with a high boiling point; a carbonate-based additive

compound having substituents selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>), and an organic sulfone-based compound of formula (3):



wherein R<sub>1</sub> and R<sub>2</sub> are independently selected from the group consisting of a primary, secondary, or tertiary alkyl group, an alkenyl group, an aryl group, and a cycloalkyl group, or R<sub>1</sub> and R<sub>2</sub> are bound together to form a cyclic ring, wherein the carbonate-based additive compound is added in an amount of 0.01 to 10 wt%.

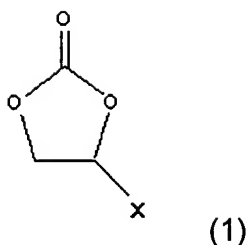
30. (ORIGINAL) The electrolyte of a lithium secondary battery according to claim 29, wherein R<sub>1</sub> and R<sub>2</sub> are independently selected from the group consisting of a C<sub>1</sub> to C<sub>4</sub> alkyl, a C<sub>2</sub> to C<sub>4</sub> alkenyl, a C<sub>6</sub> to C<sub>14</sub> aryl, and a C<sub>3</sub> to C<sub>6</sub> cycloalkyl.

31. (ORIGINAL) The electrolyte of a lithium secondary battery according to claim 29, wherein either one of R<sub>1</sub> and R<sub>2</sub> is a vinyl.

32. (ORIGINAL) The electrolyte of a lithium secondary battery according to claim 29, wherein the sulfone-based compound is selected from the group consisting of vinyl sulfone, methyl sulfone, methylvinyl sulfone, ethylvinyl sulfone, phenyl sulfone, phenylvinyl sulfone, chlorophenylvinyl sulfone, fluorophenylvinyl sulfone, benzyl sulfone, tetramethylene sulfone, butadiene sulfone, and a mixture thereof.

33. (ORIGINAL) The electrolyte of a lithium secondary battery according to claim 29, wherein the carbonate-based additive compound is a cyclic carbonate.

34. (ORIGINAL) The electrolyte of a lithium secondary battery according to claim 29 wherein the carbonate-based additive compound is a carbonate represented by formula (1):



wherein X is selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>).

35. **(ORIGINAL)** The electrolyte of a lithium secondary battery according to claim 29 wherein the carbonate-based additive compound is fluoroethylene carbonate.

36. **(CANCELLED)**

37. **(CURRENTLY AMENDED)** The electrolyte of a lithium secondary battery according to ~~claim 36~~claim 29, wherein the carbonate-based additive compound is added in an amount of 0.01 to 5 wt% based on the total amount of the electrolyte.

38. **(CURRENTLY AMENDED)** A lithium secondary battery comprising:  
a positive electrode including a material that reversibly intercalates/deintercalates lithium ions, or a material that reversibly forms a lithium-containing compound as a positive active material;  
a negative electrode including a lithium metal, a lithium-containing alloy, or a material that reversibly intercalates/deintercalates lithium ions; and  
an electrolyte comprising:  
lithium salts;  
an organic solvent with a high boiling point; and  
a carbonate-based additive compound having substituents selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>), wherein the carbonate-based additive compound is added in an amount of 0.01 to 10 wt% based on a total amount of the electrolyte.

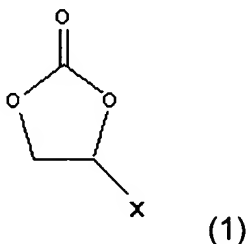
39. **(ORIGINAL)** The lithium secondary battery according to claim 38, wherein the positive electrode includes a lithium-nickel-based or a lithium-nickel-manganese-based oxide.

40. **(ORIGINAL)** The lithium secondary battery according to claim 38, wherein the negative electrode includes graphite.

41. **(ORIGINAL)** The lithium secondary battery according to claim 40, wherein the graphite comprises 1 to 100 wt% of natural graphite.

42. **(ORIGINAL)** The lithium secondary battery according to claim 38, wherein the lithium secondary battery includes a lithium ion battery or a lithium polymer battery.

43. **(WITHDRAWN)** An electrolyte of a lithium secondary battery comprising:  
a mixed solvent formed by mixing  $\gamma$ -butyrolactone (GBL)/ethylene carbonate (EC)/ethyl methyl carbonate (EMC)/dimethyl carbonate (DMC)/fluorobenzene (FB),  $\text{LiPF}_6$  to a concentration of 0.6 to 2.0 M, an ethylene carbonate derivative added in an amount to obtain 0.01 to 10 wt%, based on a total amount of the electrolyte, and fluoroethylene carbonate,  
wherein the ethylene carbonate derivative includes electron-withdrawing groups with high electronegativity selected from the group consisting of a halogen, a cyano (CN), and a nitro ( $\text{NO}_2$ ), the ethylene carbonate derivative being represented by formula (1):



wherein X is selected from the group consisting of a halogen, a cyano (CN), and a nitro ( $\text{NO}_2$ ).

44. **(WITHDRAWN)** The electrolyte of claim 43, wherein the electrolyte is included in a lithium secondary battery, the lithium battery comprising:  
a positive electrode prepared by:  
mixing in N-methyl-2-pyrrolidone (NMP) to prepare a positive material slurry:  
a positive active material comprising  $\text{LiCoO}_2$ ;

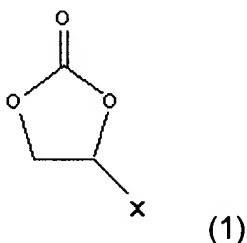


a conductive agent; and  
a binder comprising polyvinylidene fluoride (PVdF);  
coating, on an aluminum foil;  
drying;  
compressing; and  
a negative electrode formed by:  
mixing to prepare a negative material slurry:  
a negative active material comprising artificial graphite;  
oxalic acid; and  
a binder comprising PVdF;  
coating on a copper foil;  
drying; and  
compressing;  
a separator comprising a polyethylene porous film separator interposed between the positive electrode and the negative electrode,  
wherein the combination of the positive electrode, separator and the negative electrode are wound together and placed in a prismatic can to form the battery.

45. **(WITHDRAWN)** A method of preparing an electrolyte of a lithium secondary battery, comprising:

mixing  $\gamma$ -butyrolactone (GBL)/ethylene carbonate (EC)/ethyl methyl carbonate (EMC)/dimethyl carbonate (DMC)/fluorobenzene (FB) to prepare an organic mixed solvent;  
adding  $\text{LiPF}_6$  in an amount to form a concentration of 0.6 to 2.0 M  $\text{LiPF}_6$ ; and  
adding an ethylene carbonate derivative,

wherein the ethylene carbonate derivative includes electron-withdrawing groups with high electronegativity selected from the group consisting of a halogen, a cyano (CN), and a nitro ( $\text{NO}_2$ ), the ethylene carbonate derivative being represented by formula (1):



wherein X is selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>).

46. **(WITHDRAWN)** A method of preparing a secondary lithium battery, comprising:

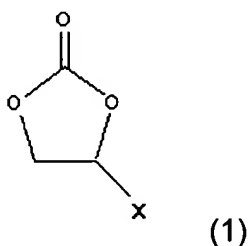
forming an electrolyte by:

mixing  $\gamma$ -butyrolactone (GBL)/ethylene carbonate (EC)/ethyl methyl carbonate (EMC)/dimethyl carbonate (DMC)/fluorobenzene (FB) to prepare an organic mixed solvent;

adding LiPF<sub>6</sub> in an amount to form a concentration of 0.6 to 2.0 M LiPF<sub>6</sub>;

adding an ethylene carbonate derivative,

wherein the ethylene carbonate derivative includes electron-withdrawing groups with high electronegativity selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>), the ethylene carbonate derivative being represented by formula (1):



wherein X is selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>);

preparing a positive electrode by:

mixing in N-methyl-2-pyrrolidone (NMP) to prepare a positive material slurry:

a positive active material comprising LiCoO<sub>2</sub>;

a conductive agent; and

a binder comprising polyvinylidene fluoride (PVdF);

coating the positive material slurry on an aluminum foil to form a first coated foil;

drying the first coated foil;

compressing the first coated foil; and

forming a negative electrode by:

mixing to prepare a negative material slurry:

a negative active material comprising artificial graphite;

oxalic acid; and

a binder comprising PVdF;

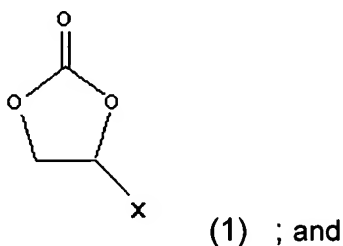
coating the negative material slurry on a copper foil to form a second coated foil;  
 drying the second coated foil; and  
 compressing the second coated foil;  
 interposing a separator comprising a polyethylene porous film separator between the positive electrode and the negative electrode,  
 winding the combination of the positive electrode, separator and the negative electrode together to form a wound combination;  
 placing the wound combination in a prismatic can; and  
 injecting the electrolyte into the can to form the battery.

47. **(WITHDRAWN)** An electrolyte of a lithium secondary battery comprising:  
 a mixture comprising:

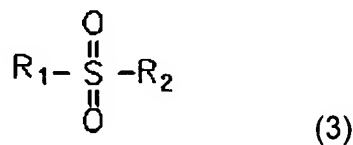
$\gamma$ -butyrolactone (GBL);

lithium salts;

a carbonate-based additive compound represented by formula (1) wherein X is selected from the group consisting of a halogen, a cyano (CN), and a nitro (NO<sub>2</sub>):



an organic sulfone-based compound of formula (3):



wherein R<sub>1</sub> and R<sub>2</sub> are independently selected from the group consisting of a primary, secondary or tertiary alkyl group, an alkenyl group, an aryl group, and a cycloalkyl group, or R<sub>1</sub> and R<sub>2</sub> are bound together to form a cyclic ring.

48. **(WITHDRAWN)** The electrolyte according to claim 47, wherein the

carbonate-based additive compound is fluoroethylene carbonate.

49. **(WITHDRAWN)** The electrolyte according to claim 47, wherein the carbonate-based additive compound is added in substantially an amount of 0.01 to 10 wt% based on a total weight of the electrolyte.

50. **(WITHDRAWN)** The electrolyte according to claim 49, wherein the carbonate-based additive compound is added in substantially an amount of 0.01 to 5 wt% based on the total weight of the electrolyte.

51. **(WITHDRAWN)** The electrolyte according to claim 47, wherein the lithium salts are at least one selected from the group consisting of  $\text{LiPF}_6$ ,  $\text{LiBF}_4$ ,  $\text{LiSbF}_6$ ,  $\text{LiAsF}_6$ ,  $\text{LiClO}_4$ ,  $\text{LiCF}_3\text{SO}_3$ ,  $\text{Li}(\text{CF}_3\text{SO}_2)_2\text{N}$ ,  $\text{LiC}_4\text{F}_9\text{SO}_3$ ,  $\text{LiSbF}_6$ ,  $\text{LiAlO}_4$ ,  $\text{LiAlCl}_4$ ,  $\text{LiN}(\text{C}_x\text{F}_{2x+1}\text{SO}_2)(\text{C}_y\text{F}_{2y+1}\text{SO}_2)$  (where  $x$  and  $y$  are natural numbers),  $\text{LiCl}$ , and  $\text{LiI}$ .

52. **(WITHDRAWN)** The electrolyte according to claim 47, wherein  $\text{R}_1$  and  $\text{R}_2$  are independently selected from the group consisting of a  $\text{C}_1$  to  $\text{C}_4$  alkyl, a  $\text{C}_2$  to  $\text{C}_4$  alkenyl, a  $\text{C}_6$  to  $\text{C}_{14}$  aryl, and a  $\text{C}_3$  to  $\text{C}_6$  cycloalkyl.

53. **(WITHDRAWN)** The electrolyte according to claim 47, wherein  $\text{R}_1$  and  $\text{R}_2$  are independently a halogen-substituted alkyl group, an alkenyl group, an aryl group, or a cycloalkyl group.

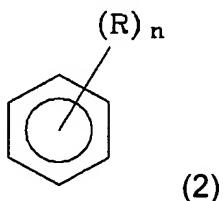
54. **(WITHDRAWN)** The electrolyte according to claim 47, wherein the sulfone-based compound is selected from the group consisting of vinyl sulfone, methyl sulfone, methylvinyl sulfone, ethylvinyl sulfone, phenyl sulfone, phenylvinyl sulfone, chlorophenylvinyl sulfone, fluorophenylvinyl sulfone, benzyl sulfone, tetramethylene sulfone, butadiene sulfone, and a mixture thereof.

55. **(WITHDRAWN)** The electrolyte according to claim 47, wherein the sulfone-based compound is added in substantially an amount of 0.01 to 10 wt% based on the total weight of the electrolyte.

56. **(WITHDRAWN)** The electrolyte according to claim 55, wherein the sulfone-based compound is added in substantially an amount of 0.01 to 6 wt% based on the total amount of the electrolyte.

57. **(WITHDRAWN)** The electrolyte according to claim 51, wherein the lithium salts are used in a concentration ranging from 0.6 to 2.0 M.

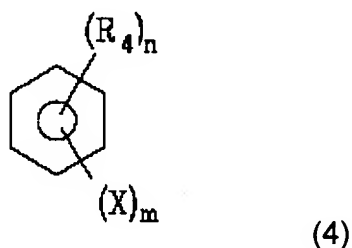
58. **(WITHDRAWN)** The electrolyte according to claim 47, wherein the electrolyte further comprises an aromatic hydrocarbon organic solvent of formula (2):



wherein R is a halogen or a  $C_1$  to  $C_{10}$  alkyl, and n is an integer of 0 to 6.

59. **(WITHDRAWN)** The electrolyte according to claim 58, wherein the aromatic hydrocarbon organic solvent is selected from the group consisting of benzene, fluorobenzene, toluene, fluorotoluene, trifluorotoluene, xylene, and a mixture thereof.

60. **(WITHDRAWN)** The electrolyte according to claim 47, wherein the electrolyte further comprises a compound of formula (4):



wherein  $R_4$  is a  $C_1$  to  $C_{10}$  alkyl, a  $C_1$  to  $C_{10}$  alkoxy, or a  $C_6$  to  $C_{10}$  aryl, X is a halogen, and m and n are integers ranging from 1 to 5, where  $m+n$  is less than or equal to 6.

61. **(WITHDRAWN)** The electrolyte according to claim 60, wherein the compound of formula (4) is a compound selected from the group consisting of 3-fluoroanisole, 3-chloroanisole, 3-bromoanisole, 4-fluoroanisole, 4-chloroanisole, 4-bromoanisole, 2,4-difluoroanisole, 3,5-difluoroanisole, 3-chloro-5-fluoroanisole, and a mixture thereof.

62. **(WITHDRAWN)** The electrolyte according to claim 60, wherein the compound of formula (4) is added in substantially an amount of 0.01 to 10 wt% based on the total amount of the electrolyte.

63. **(WITHDRAWN)** The electrolyte according to claim 62, wherein the compound of formula (4) is added in substantially an amount of 0.01 to 6 wt% based on the total amount of the electrolyte.